## Diatomic and Polyatomic elements to MEMORIZE

Monoatomic elements (exist on their own)	$C_{(s)}$ noble gases all metals
Diatomic molecular elements (exist as pairs)	$H_{2(g)}  N_{2(g)}  Cl_{2(g)}  F_{2(g)}  Br_{2(g)}  I_{2(s)}  O_{2(g)}$
Polyatomic molecular elements (exist in groups larger that 2)	$\mathbf{P}_{4(s)}$ $\mathbf{S}_{8(s)}$